

STRUCTURE OF CYTOSININE, A STRUCTURAL COMPONENT OF BLASTICIDIN S

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In previous papers^{1,2}, we have reported the isolation and characterization of a new nucleoside designated cytosinine from the hydrolysis products of blasticidin S, a useful antibiotic against rice blast disease. We wish to present the details leading to the assignment of structure I to cytosinine.

Cytosinine (I), $C_{10}H_{12}O_4N_4$, m.p. 244-245°, $[\alpha]_D^{18} -20^\circ$ (c, 1 in H_2O), pKa 2.4, 4.6 and 8.0, $\lambda_{max}^{0.1N.HCl}$ 274 m μ (ϵ 13,800), $\lambda_{max}^{0.1N.NaOH}$ 267 m μ (ϵ 7,500), showed the existence of two ethylenic groups (NMR (in D_2O) ppm: 7.72 and 6.14 (2H d, AB type, J = 7.5 cps) and 6.45-6.00 (2H broad)), two amino groups (Van Slyke determination) and one carboxylic group. By the action of methanol containing 3% hydrogen chloride, I gave corresponding methyl ester dihydrochloride, $C_{10}H_9O_3N_4(OCH_3) \cdot 2HCl$,

m.p. 220-223°(dec.), IR (Nujol) 1730 and 1230 cm^{-1} , pKa 4.6 and 8.0, which was acetylated with acetic anhydride and triethylamine to afford N, N'-diacetyl methylester (III), $\text{C}_{10}\text{H}_9\text{O}_3\text{N}_4 \cdot (\text{OCH}_3) \cdot (\text{CH}_3\text{CO})_2$, m.p. 263-266°.

UV spectral evidence indicated probable presence of cytosine nucleus, on which a substituent attached at N-1 position.

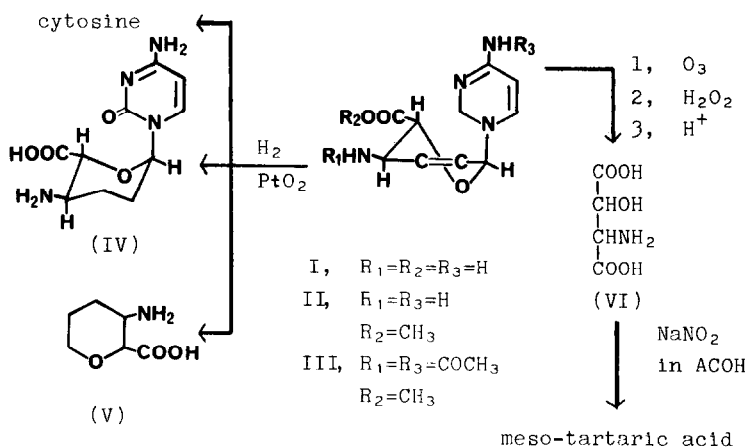
Acid hydrolysis of I gave one mole each of cytosine, levulinic acid, ammonia and carbon dioxide as characterizable products, whereas further attempt to cleave the glycosidic linkage of I was unsuccessful without extensive decomposition.

In an effort to gain information regarding the sugar portion^{*1}, reduction and oxidative cleavage of ethylenic function in I were undertaken. On reduction with PtO_2 in acetic acid, I gave a mixture of the dihydrocytosine (IV), $\text{C}_{10}\text{H}_{14}\text{O}_4\text{N}_4$, m.p. 227-229°(dec.), NMR (D_2O) ppm: 7.25 and 5.45 (2H d, $J = 7.5$ cps), 5.10 (1H m), 3.50 (1H d, $J = 10.5$ cps), 2.70 (1H m) and 1.40 (6H m), hydrogenolysis product (V), $\text{C}_6\text{H}_{11}\text{O}_3\text{N}$, m.p. 275-278°, and cytosine. The facile hydrogenolysis of I to cytosine and V indicated that an ethylenic function in C_6 moiety located at allylic position to the glycosidic C-N bond.

*1 The presence of sugar moiety was presumed, since I gave levulinic acid on acid hydrolysis.

V characterized as monobasic monoacidic compound of pKa 9.2 and 2.4 gave corresponding N-acetyl methylester, $C_6H_9O_2N \cdot (OCH_3) \cdot (CH_3CO)$, m.p. 130° , IR (Nujol) 1740, 1670 and 1230 cm^{-1} . The IR spectra of V and deuterated V exhibiting characteristic bands of tetrahydropyran³ at 1100 and 1060 cm^{-1} (C-O-C stretching vibration) established the presence of tetrahydropyran ring.

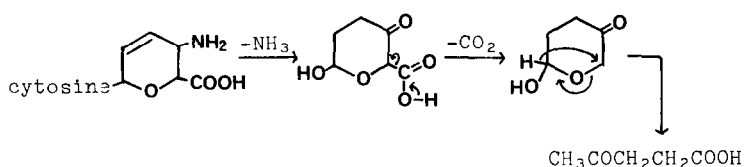
Ozonolysis of III in acetic acid followed by oxidation with hydrogen peroxide and acid hydrolysis furnished β -hydroxy aspartic acid (VI), $C_4H_7O_5N$, m.p. $287\text{-}288^\circ(\text{dec.})$, $[\alpha]_D^{20} -47^\circ$ (c, 1 in H_2O). On diazotization, VI afforded meso-tartaric acid, therefore, the erythro-D-configuration⁴ was established. VI was a substantially important structural fragment of the sugar portion in I and was also obtained from blasticidin S itself by a similar ozonolysis.



These results established the structure of V as 3-amino-tetrahydropyran-2-carboxylic acid. On the basis of these accumulated evidence the structure of cytosinine was elucidated as Formula I.

Stereochemical feature of I was next to be considered, e.g. the configuration of erythro-D- β -hydroxy aspartic acid provided evidence for the (R)-configuration at C₄' and C₅' respectively. The NMR spectrum of IV showing a distinct doublet at 3.50 ppm (1H, J = 10.5 cps) assigned for C₅' proton established the C₄'-C₅' protons to be diaxial. The configuration of glycosidic linkage remained uncertain, but it was presumed to be β form, since nucleosides from natural source were reported as β .

It was necessary to consider the mechanism of levulinic acid formation from I. Transformation of hexoses into levulinic acid by the action of acids was well established⁵. Similarly, the formation of levulinic acid, ammonia and carbon dioxide from I could be explained as follows:



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